

**DREAMERS EDU HUB**  
**PRE-BOARD EXAMINATION PAPER 2025-26**

**CLASS 10<sup>th</sup>**

**SCIENCE**

**18.12.2025**

Time allowed : 3 hours

Maximum Marks : 70

General Instructions:

Read the following instructions very carefully and strictly follow them:

- I) This question paper comprises 39 questions. All questions are compulsory.
- ii) This question paper is divided into FIVE sections viz. Section A, B, C, D and. E.
- iii) In Section A question number 1 to 20 are Multiple Choice Questions (MCQs) carrying 1 mark each.
- iv) In Section B question number 21 to 26 are Very Short Answer (VSA) type questions carrying 2 marks each. Answer to these questions should be in the range of 30 to 50 words.
- v) In Section C question number 27 to 33 are Short Answer (SA) type questions carrying 3 marks each. Answer to these questions should be in the range of 50 to 80 words.
- vi) In Section D question number 34 to 36 are Long Answer (LA) type questions carrying 5 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii) In Section E question number 37 to 39 are of 3 source-based/case-based units of assessment carrying 4 marks each with sub-parts.
- viii) There is no overall choice. However, an internal choice has been provided in some Sections. Only one of the alternatives has to be attempted in such questions.

**SECTION-A**

Select and write one most appropriate option out of the four options given for each of the questions 1 to 20:

1. Consider the following statements about homologous series of carbon compounds: (1)  
a) All succeeding members differ by –  $\text{CH}_2$  unit.  
b) Melting point and boiling point increases with increasing molecular mass.  
c) The difference in molecular masses between two successive members is 16 u.  
d)  $\text{C}_2\text{H}_3$  and  $\text{C}_3\text{H}_4$  are NOT the successive members of alkyne series.  
The correct statements are –  
A.(a) and (b)      B.(b) and (c)      C.(a) and (c)      D.(c) and (d)
2. The number of shells required to write the electronic configuration of Potassium (At. No. 19) (1)  
a) 1.      b) 2      c) 3.      d) 4
3. Select from the following a process in which a combination reaction is involved: (1)  
a) Black and White photography      b) Burning of coal      c) Burning of methane      d) Digestion of food
4. The oxide which can react with HCl as well as KOH to give corresponding salt and water is (1)  
a) CuO.      b)  $\text{Na}_2\text{O}$       c)  $\text{Al}_2\text{O}_3$ .      d)  $\text{K}_2\text{O}$
5. Which of the following is an alloy of copper and tin? (1)  
a) Nichrome.      b) Brass      c) Constantan.      d) Bronze
6. Tooth decay begins at the pH of (1)  
a) 5.1.      b) 5.8      c) 6.5.      d) 8.0
7. Solid Calcium oxide reacts vigorously with water to form Calcium hydroxide accompanied by the liberation of heat. From the information given above it may be concluded that this reaction (1)  
a) Is endothermic and pH of the solution formed is more than 7.      b) Is exothermic and pH of the solution formed is 7.  
c) Is endothermic and pH of the solution formed is 7.      d) Is exothermic and pH of the solution formed is more than 7.
8. In human respiratory system, when a person breathes in, the position of ribs and diaphragm will be 1) (1)  
a) Lifted ribs and curve/dome shaped diaphragm.      b) Lifted ribs and flattened diaphragm.  
c) Relaxed ribs and flattened diaphragm.      d) Relaxed ribs and curve/dome shaped diaphragm.
9. Select out of the following a gland which does NOT occur as a pair in the human body (1)  
a) Pituitary.      b) Testis      c) Ovary.      d) Adrenal

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(1)

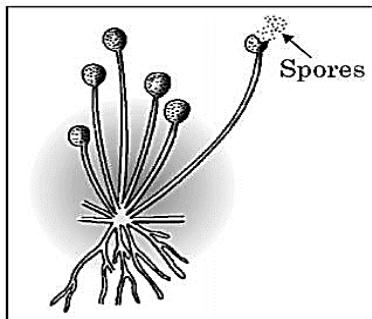
10. Which of the following statement(s) is (are) true about human heart?

- a) Right atrium receives oxygenated blood from lungs through pulmonary artery.
- b) Left atrium transfers oxygenated blood to left ventricle which sends it to various parts of the body.
- c) Right atrium receives deoxygenated blood through vena cava from upper and lower body.
- d) Left atrium transfers oxygenated blood to aorta which sends it to different parts of the body.

A)(a).    B) (a) and (d)    C) (b) and (c).    D)(b) and (d)

11. Which one of the following organism is represented by this diagram?

(1)



(A) Spirogyra.    (B) Planaria    (C) Yeast    (D) Rhizopus

12. A cross made between two pea plants produces 50% tall and 50% short pea plants. The gene combination of the parental pea plants must be

(1)

- a) Tt and Tt
- b) TT and Tt
- c) Tt and tt.
- d) TT and tt

13. Strength of magnetic field produced by a current carrying solenoid DOES NOT depend upon

(1)

- a) Number of turns in the solenoid
- b) Direction of the current flowing through it
- c) radius of solenoid
- d) Material of core of the solenoid

14. S.I. unit of electrical resistivity is

(1)

- a) Ohm per metre<sup>2</sup>
- b) Ohm per metre<sup>3</sup>
- c) ohm metre
- d) Ohm metre<sup>3</sup>

15. The minimum resistance which can be made using five resistors each of resistance  $10\ \Omega$  is

(1)

- a)  $1/50\Omega$
- b)  $1/5\Omega$
- c)  $2\ \Omega$ .
- d)  $1\ \Omega$

16. Consider the following statements in the context of human eye

(1)

- a) The diameter of the eye ball is about 2.3 cm.
- b) Iris is a dark muscular diaphragm that controls the size of the pupil.
- c) Most of the refraction for the light rays entering the eye occurs at the crystalline lens.
- d) While focusing on the objects at different distances the distance between the crystalline lens and the retina is adjusted by ciliary muscles.

The correct statements are –

- A)(a) and (b).    B)(a), (b) and (c)
- C) (b), (c) and (d).    D)(a), (c) and (d)

17 to 20 are Assertion – Reason based questions:

These questions consist of two statements Assertion (A) and Reason <sup>®</sup>. Answer these questions selecting the appropriate option given below:

- A. Both (A) and (R) are true and (R) is the correct explanation of (A).
- B. Both (A) and (R) are true, but (R) is not the correct explanation of (A).
- C.(A) is true, but R is false.
- D.(A) is false, but R is true.

17. **Assertion (A):** The deflection of a compass needle placed near a current carrying wire decreases when the magnitude of an electric current in the wire is increased.

(1)

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**Reason(R):-** Strength of the magnetic field at a point due to a current carrying conductor increases on increasing the current in the conductor.

**18. Assertion (A):** Human female has a perfect pair of sex chromosome. (1)

**Reason(R):-** Sex chromosome contributed by the human male in the zygote decides the sex of a child.

**19. Assertion (A):** Myopic eye cannot see distant objects distinctly. (1)

**Reason(R):-** For the correction of myopia converging lenses of appropriate power are prescribed by eye-surgeons.

**20. Assertion (A):** Metals in the middle of activity series are found in nature as sulphides or carbonates. (1)

**Reason(R):-** The sulphide ores are calcinated whereas carbonate ores are roasted to extract metals from them.

**SECTION-B**

**Q. Nos. 21 to 26 are very short answer questions.**

21. (a) Define a decomposition reaction. Write an equation to show thermal decomposition of ferrous sulphate crystals. (2)

**OR**

(b) What is meant by a balanced chemical equation? Why is it necessary for the equation to be balanced? (2)

22. Two test tubes A and B are taken, each containing one mL of starch solution. Add 1 mL of saliva to test tube 'A' only and leave both the test tubes undisturbed for a few minutes. Now add a few drops of dilute iodine solution to both the test tubes.

a) Which one of the two test tubes shows change in colour? Write the changed colour observed in this test tube. (2)

b) What can we conclude from this activity?

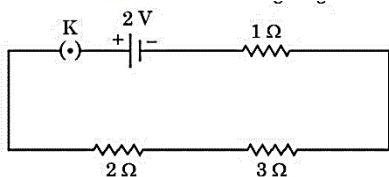
23. Name two types of germ cells present in human beings. List two structural differences between the two. (2)

24. (a) State two laws of refraction of light. (2)

**OR**

a) Define the term absolute refractive index of a medium. A ray of light enters from vacuum to glass of absolute refractive index 1.5. Find the speed of light in glass. The speed of light in vacuum is  $3 \times 10^8$  m/s. (2)

25. Use Ohm's law to determine the potential difference across the  $3\Omega$  resistor in the circuit shown in the following diagram when key is closed: (2)



26. Name the term used for the materials which cannot be broken down by biological processes. Give two ways by which they harm various components of an ecosystem.

**SECTION-C**

**Q. Nos. 27 to 33 are short answer questions.**

27. (a) Give reasons for the following:

i) Alveoli in lungs are richly supplied with blood capillaries.

ii) Respiratory pigment in the blood takes up oxygen and not carbon dioxide. (3)

iii) During anaerobic respiration, a 3-carbon molecule is formed as an end product instead of  $\text{CO}_2$  in human beings.

**OR**

27. (b) (i) Name the movements that occur all along the gut in human digestive system. How do they help in digestion?

(ii) Where is bile juice stored in human body? List two roles of bile juice. (3)

28. (a) In angiosperms why fertilisation cannot take place in flowers if pollination does not take place? Where is zygote located in a flower after fertilisation? What does it develop into? (3)

(b) Write the names of those parts of a flower which serve the same function as the following do in animals:

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I) Testis      (ii) ovary

29.a) State any two observations when an electric current is passed through acidulated water, in a container having each electrode covered by test tubes filled with water. (3)

b) Write the ratio of the mass of the gas collected at the cathode to the mass of the gas collected at the anode.

30. Draw a labelled diagram to show electrolytic refining of copper. State what happens when electric current is passed through the electrolyte taken in this case. (3)

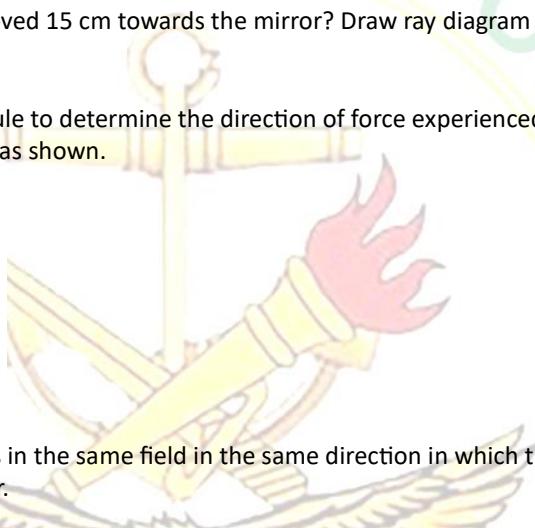
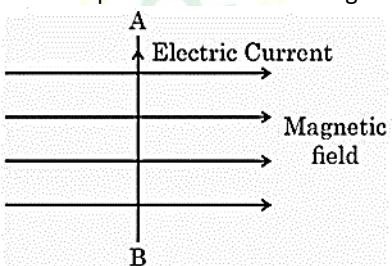
31. (a) An object is placed in front of a concave mirror of focal length 12 cm. If distance of the object from the pole of the mirror is 8 cm, then use mirror formula to determine the position of the image formed. Draw a labelled ray diagram to justify your answer in this case. (3)

**OR**

(b) (i) The image of an object formed by a mirror is real, inverted and is of magnification-1. If the image is at a distance of 30 cm from the mirror, where is the object placed? Give reason to justify your answer. (3)

(ii) Where would the image be if the object is moved 15 cm towards the mirror? Draw ray diagram for the new position of the object to justify your answer.

32.(a) State Fleming's left hand rule. Apply this rule to determine the direction of force experienced by a straight current carrying conductor AB placed in a uniform magnetic field as shown. (3)



(b) What will happen to an electron which enters in the same field in the same direction in which the current is flowing in the conductor AB? Give reason to justify your answer.

33. Use of pesticides to protect our crops affect organisms at various trophic levels especially human beings. Name the phenomenon involved and explain how does it happen. (3)

**SECTION-D**

**Q. No. 34 to 36 are long answer questions.**

34.(a) Upper half of a convex lens is covered with a black paper. Draw a ray diagram to show the formation of image of an object placed at a distance of  $2F$  from such a lens. Mention the position and nature of the image formed. State the observable difference in the image. Obtain if the lens is uncovered. Give reason to justify your answer. (5)

(b) An object is placed at a distance of 30 cm from the optical centre of a concave lens of focal length 15 cm. Use lens formula to determine the distance of the image from the optical centre of the lens.

35. (a). (i) Give reason why carbon can neither form  $C^{++}$  cations nor  $C^{-}$  anions but form covalent compounds.

(ii) What is homologous series of carbon compound? Write the molecular formula of any two consecutive members of homologous series of aldehydes.

(iii) Draw the structure of the molecule of cyclohexane ( $C_6H_{12}$ ). (5)

**OR**

35. (b). (i) Name a commercially important carbon compound having functional group-OH and write its molecular formula.

(ii) Write chemical equation to show its reaction with.

1. Sodium metal

2. Excess conc. Sulphuric acid

3. Ethanoic acid in the presence of an acid catalyst

4. Acidified potassium dichromate

Also write the name of the product formed in each case.

